# Five New Diterpenoids from Pseudolarix kaempferi 

Sheng-Ping Yang, Yan Wu, and Jian-Min Yue*<br>Institute of Materia Medica, Shanghai Institutes for Biol ogical Sciences, Chinese Academy of Sciences, 294 Taiyuan Road, Shanghai, 200031, People's Republic of China

Received J anuary 11, 2002


#### Abstract

Five new diterpenoids, pseudolaric acids F (1), G (2), and H (3), 2',3'-dihydroxy-1'-propoxypseudolarate B (4), and 6'-O-acetylpseudolaric acid B O- $\beta$-D-glucopyranoside (5), al ong with nine known diterpenoids, pseudolaric acids $\mathrm{A}, \mathrm{B}$, and $\mathrm{C}_{1}$, deacetylpseudolaric acid $\mathrm{C}_{2}$, deacetylpseudolaric acid A , methyl pseudolarate A , methyl pseudolarate B , pseudolaric acid $\mathrm{A}-\mathrm{O}-\beta$-D-glucopyranoside, and pseudolaric acid B-O- $\beta$-D-glucopyranoside, were isolated from the root bark of Pseudolarix kaempferi. Their structures and stereochemistry were elucidated mainly by spectral data, especially 2D NMR techniques.


A series of novel diterpenoids ${ }^{1}$ and triterpenoids ${ }^{2}$ have been isolated from the root bark of Pseudol arix kaempferi Gordon (Pinaceae), known as "Tu-J in-Pi" in traditional Chinese medicine and used for the treatment of skin diseases caused by fungal infections. ${ }^{1 a}$ A number of triterpenoids ${ }^{3}$ have been isolated from the fruits of this plant. Some of the compounds obtained from this species, such as pseudolaric acids A and B, pseudolarolide B, and isopseudolarifuroic acids A and B, have shown potent cytotoxic ${ }^{4}$ and antimicrobial ${ }^{1 b, 2 b, 5}$ potencies. In the present investigation, five new diterpenoids, namely, pseudolaric acids $F$ (1), G (2), and $H$ (3), 2', $3^{\prime}$-dihydroxy-1'-propoxypseudolarate $B$ (4), and 6'-O-acetylpseudolaric acid B-O-$\beta$-D-glucopyranoside (5), along with nine known diterpenoids, pseudolaric acids $A, B$, and $C_{1}$, deacetylpseudolaric acid $C_{2}$, deacetylpseudolaric acid $A$, methyl pseudolarate A , methyl pseudolarate B , pseudolaric acid $\mathrm{A}-\mathrm{O}-\beta$-D-glucopyranoside, and pseudolaric acid B-O- $\beta$-D-glucopyranoside, were isolated from the root bark of Pseudolarix kaempferi. The structures and stereochemistry of 1-5 were elucidated by spectral data interpretation. Deacetylpseudolaric $A$, methyl pseudolarate $B$, and methyl pseudoIarate A were isolated from the root bark of $P$. kaempferi for the first time. ${ }^{1,6}$

Pseudolaric acid F (1), with the molecular formula $\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{O}_{7}$ as determined by HREIMS (m/z 384.1571 [M $\left.\mathrm{H}_{2} \mathrm{O}\right]^{+}$, calcd 384.1573), was obtained as a white amorphous powder. The ${ }^{1} \mathrm{H}$ NMR (Table 1) and ${ }^{13} \mathrm{C}$ NMR (Table 2) data showed the presence of four methyls, four methylenes, five methines, and nine quaternary carbons. The ${ }^{1} \mathrm{H} N M R,{ }^{13} \mathrm{C}$ NMR (DEPT), and HMQC spectra permitted the characterization of one disubstituted and two trisubstituted double bonds, one ketone group, three ester carbonyls, and two oxygenated quaternary carbons. Four proton signals (each 3 H ) at $\delta 1.64$ (s), 1.73 (d), 1.89 (d), and 2.02 (s) in the ${ }^{1} \mathrm{H}$ NMR spectrum were assigned to four methyl groups. The singlet methyl group at $\delta 2.02$ that correlated with one of the carbonyl groups in the HMBC spectrum showed the occurrence of an acetyl group. These data indi cated that compound $\mathbf{1}$ was most likely an analogue of pseudolaric acid $A$, which also was obtained in the present investigation. Analysis of the ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY, HMQC, and HMBC confirmed that $\mathbf{1}$ has the same basic skeleton as pseudolaric $\operatorname{acid} A$. The absence of a molecular ion and the presence of a weak characteristic ion at $\mathrm{m} / \mathrm{z} 342$ [ $\mathrm{M}-\mathrm{AcOH}$ ] in the EIMS, a typical observation for the derivatives of pseudo-

[^0]
1

2

3



Iaric acid $A,{ }^{7}$ further supported this condusion. Comparison of the ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of compound $\mathbf{1}$ with those of pseudolaric acid $A^{1 a, b}$ suggested that the two compounds possess the same side chain.

The oxygenated quaternary carbon signal at $\delta 84.8$ was assigned unambiguously to $\mathrm{C}-11$ in $\mathbf{1}$ from the correlation with the proton signal of $\mathrm{H}-12$ at $\delta 1.64$ in the HMBC spectrum. Consequently, the other oxygenated quaternary carbon signal at $\delta 88.2$ was assigned to C-4 bearing an

Table 1. ${ }^{1} \mathrm{H}$ NMR Data of Compounds $\mathbf{1 - 5} \mathbf{5}^{\mathrm{a}, \mathrm{b}}$

| proton | 1 | 2 | 3 | 4 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1.93 (m) | 1.81 (2H, m) | 1.99 (m) | 1.80 (2H, m) | 1.91 (2H, m) |
|  | 1.81 (m) |  | 1.92 (m) |  |  |
| 2 | 1.88 (2H, m) | 1.79 (2H, m) | 2.02 (m) | 1.80 (2H, m) | 1.87 (2H, m) |
|  |  |  | 1.92 (m) |  |  |
| 3 | 3.27 (d, 5.0) | 3.14 (br d, 4.4) | 3.19 (d, 6.2) | 3.31 (m) | 3.40 (br d, 3.0) |
| 45 |  |  |  |  |  |
|  | 3.45 (ddd, 17.3, 7.0, 1.0) | 3.41 (br d, 14.0) | 3.55 (dd, 16.4, 11.2) | 3.05 (dd, 13.7, 5.8) | 3.21 (dd, 14.1, 5.6) |
|  | 3.07 (ddd, 17.3, 4.9, 1.6) | 2.34 (dd, 14.0, 4.8) | 2.56 (dd, 16.4, 2.7) | 1.67 (m) | 1.65 (m) |
| 6 | 6.09 (m) | 5.12 (m) | 5.38 (ddd, 11.2, 2.7, 1.1) | 2.66 (2H, br d, 6.6) | 3.02 (dd, 15.4, 5.6) |
|  |  |  |  |  | $2.28 \text { (m) }$ |
| 7 |  |  |  |  |  |
| 8 |  | 4.46 (br d, 10.9) | 5.74 (dd, 12.3, 1.1) | 7.09 (m) | 7.36 (m) |
| 9 | 3.04 (d, 16.0) | 2.26 (br d, 13.5) | 5.70 (d, 12.3) | 2.79 (dd, 15.6, 6.3) | 2.81 (dd, 15.2, 9.5) |
|  | 2.88 (d, 16.0) | 1.85 (m) |  | 2.19 (m) | 2.63 (m) |
| 10 |  |  |  |  |  |
| 11 |  |  |  |  |  |
| 12 | 1.64 (s) | 1.59 (s) | 1.61 (s) | 1.57 (s) | 1.50 (s) |
| 13 | 6.22 (d, 15.2) | 6.18 (d, 15.1) | 6.20 (d, 15.2) | 6.22 (d, 15.1) | 5.76 (d, 15.0) |
| 14 | 6.56 (dd, 15.2, 11.4) | 6.56 (dd, 15.1, 11.5) | 6.57 (dd, 15.2, 11.8) | 6.53 (dd, 15.1, 11.5) | 6.59 (dd, 15.0, 11.9) |
| 15 | 7.19 (d, 11.4) | 7.18 (d, 11.5) | 7.19 (d, 11.8) | 7.24 (d, 11.5) | 7.41 (d, 11.9) |
| 16 ( 16 |  |  |  |  |  |
| 17 | 1.89 (d, 1.2) | 1.90 (d, 1.1) | 1.90 (d, 1.1) | 1.91 (d, 1.3) | 1.78 (s) |
| 18 (d, |  |  |  |  |  |
| 19 | 1.73 (d, 1.3) | 1.76 (d, 0.7) | 0.89 (m) |  |  |
| 20 |  |  |  |  |  |
| $\mathrm{CH}_{3} \mathrm{CO}-4$ | 2.02 (s) | 1.95 (s) | 1.91 (s) | 2.28 (s) | 2.14 (s) |
| $\mathrm{CH}_{3} \mathrm{O}$ |  |  |  | 3.65 (s) | 3.62 (s) |
| $1^{\prime}$ |  |  |  | 4.20 (m) | 6.45 (d, 7.3) |
|  |  |  |  | 4.11 (m) |  |
| $2^{\prime}$ |  |  |  | 3.87 (m) | 4.19-4.31 (4H, m) |
| $3{ }^{\prime}$ |  |  |  | 3.56 (2H, d, 3.2) |  |
| $4^{\prime}$$5^{\prime}$$6^{\prime}$ |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |  |  |  |
| 6 |  |  |  |  | 4.78 (dd, 16.9, 12.1) |
| $\mathrm{CH}_{3} \mathrm{CO}-6{ }^{\prime}$ |  |  |  |  | $1.83 \text { (s) }$ |

[^1]acetoxy group. The two proton signals at $\delta 3.45$ and 3.07 were assigned to the $\mathrm{H}_{2}-5$ by the correlations with the $\mathrm{C}-4$ and the carbon signal at $\delta 130.3$ (C-6) in the HMBC spectrum. The carbon signal at $\delta 33.4$ was then attributable to C-5 by the correlations with the two analogous proton signals of $\mathrm{H}_{2}-5$ in the HMQC spectrum. The ol efinic proton signal at $\delta 6.09$ was assigned to $\mathrm{H}-6$ by the HMBC spectral correlation with C-5 and C-4. In turn, the olefinic carbon signal at $\delta 130.3$ was assigned to C-6 by the correlation with H-6 in the HMQC spectrum, with the $\mathrm{H}-6$ signal al so having a correlation with the carbon signal at $\delta 141.3$ assignable to $\mathrm{C}-7$, indicating the presence of a double bond between $\mathrm{C}-6$ and $\mathrm{C}-7$. A quaternary carbon signal at $\mathrm{C}-10$ at $\delta 55.2$ could also be distinguished. The proton signals of $\mathrm{H}-1, \mathrm{H}-2$, and $\mathrm{H}-3$ were identified by the correlations in the ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY spectrum and confirmed by correlations between $\mathrm{H}-1$ and $\mathrm{C}-10$ and between $\mathrm{H}-3$ and $\mathrm{C}-4$ in the HMBC spectrum, and in consequence, the assignments for C-1, C-2, and C-3 were made on the basis of HMQC spectral correlations with $\mathrm{H}-1, \mathrm{H}-2$, and $\mathrm{H}-3$, respectively. The proton signals at $\delta 3.04$ and 2.88 were assigned to the $\mathrm{H}_{2}-9$ protons by the correlations with $\mathrm{C}-10$ in the HMBC spectrum. The carbon signal at $\delta 201.1$ for one ketone group was allocated to $\mathrm{C}-8$ from the correlations with the $\mathrm{H}_{2}-9$ proton signals. The proton signal for a methyl group at 1.73 $\left(3 \mathrm{H}, \mathrm{d}, \mathrm{J}=1.3 \mathrm{~Hz}\right.$ ) was attributable to the $\mathrm{H}_{3}-19$ protons from the correlations with C-6, C-7, and C-8 in the HMBC spectrum. The carbonyl signal at $\delta 173.8$ was assigned to $\mathrm{C}-20$ on the basis of the correlations of $\mathrm{H}_{2}-1$ signals ( $\delta 1.93$ and 1.81) and $\mathrm{H}_{2}-9$ signals ( $\delta 3.04$ and 2.88 ). Therefore, the structure of pseudolaric acid F was elucidated as $\mathbf{1}$.

Table 2. ${ }^{13} \mathrm{C}$ NMR Data of Compounds $\mathbf{1 - 5} \mathbf{5}^{\mathrm{a}, \mathrm{b}}$

| carbon | $\mathbf{1}$ | $\mathbf{2}$ | $\mathbf{3}$ | $\mathbf{4}$ | $\mathbf{5}$ |
| :--- | ---: | ---: | ---: | ---: | ---: |
| 1 | 33.6 | 34.7 | 33.1 | 33.8 | 33.2 |
| 2 | 25.5 | 24.2 | 27.0 | 24.8 | 24.4 |
| 3 | 49.3 | 49.1 | 49.2 | 50.1 | 49.3 |
| 4 | 88.2 | 84.5 | 87.2 | 90.3 | 89.6 |
| 5 | 33.4 | 31.7 | 33.6 | 31.4 | 30.7 |
| 6 | 130.3 | 115.1 | 121.7 | 28.2 | 20.2 |
| 7 | 141.3 | 147.9 | 135.0 | 135.2 | 134.5 |
| 8 | 201.1 | 68.4 | 128.3 | 142.7 | 141.9 |
| 9 | 44.1 | 37.7 | 128.7 | 20.5 | 27.8 |
| 10 | 55.2 | 57.5 | 60.3 | 56.0 | 55.2 |
| 11 | 84.8 | 84.4 | 84.6 | 84.4 | 83.4 |
| 12 | 28.1 | 28.0 | 28.2 | 28.5 | 27.9 |
| 13 | 144.5 | 145.2 | 144.9 | 145.5 | 145.0 |
| 14 | 122.5 | 122.3 | 122.3 | 121.9 | 120.9 |
| 15 | 137.3 | 137.5 | 137.3 | 137.8 | 138.2 |
| 16 | 129.3 | 129.1 | 129.3 | 128.6 | 127.4 |
| 17 | 12.8 | 12.8 | 12.8 | 20.6 | 12.2 |
| 18 | 169.8 | 169.8 | 169.8 | 168.5 | 166.7 |
| 19 | 20.5 | 19.4 | 25.9 | 168.4 | 167.6 |
| $1^{2}$ | 173.8 | 174.2 | 173.2 | 173.5 | 172.7 |
| 20 |  |  |  |  |  |
| $\mathrm{CH}_{3} \mathrm{CO}-4$ | 170.2 | 170.2 | 169.9 | 170.3 | 169.3 |
| $\mathrm{CH}_{3} \mathrm{CO}-4$ | 21.4 | 21.3 | 21.4 | 21.6 | 21.1 |
| $\mathrm{CH}_{3} \mathrm{O}$ |  |  |  | 52.1 | 51.5 |
| $1^{\prime}$ |  |  |  | 66.7 | 95.8 |
| $2^{\prime}$ |  |  |  | 70.6 | 73.8 |
| $3^{\prime}$ |  |  |  | 63.7 | 77.8 |
| $4^{\prime}$ |  |  |  |  | 70.6 |
| $5^{\prime}$ |  |  |  |  | 75.8 |
| $6^{\prime}$ | $\mathrm{CH}_{3} \mathrm{CO}-6^{\prime}$ |  |  |  |  |
| $\mathrm{CH}_{3} \mathrm{CO}-6^{\prime}$ |  |  |  |  | 170.4 |

[^2]Pseudolaric acid G (2) exhibited the molecular formula $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{O}_{7}$ as determined by HREIMS (m/z 386.1764 [M $\left.\mathrm{H}_{2} \mathrm{O}\right]^{+}$, calcd 386.1729) combined with ESIMS (m/z 427 [M $+\mathrm{Na}]^{+}$). The molecular composition showed that compound 2 had two more hydrogen atoms than 1. The ${ }^{1} \mathrm{H}$ NMR (Table 1) and ${ }^{13} \mathrm{C}$ NMR (Table 2) data showed the presence of four methyls, four methylenes, six methines, and eight quaternary carbons. Comparison of the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data of $\mathbf{2}$ with those of $\mathbf{1}$ suggested that both compounds have very similar structures. This was supported by the absence of a molecular ion and the presence of a weak characteristic ion at $\mathrm{m} / \mathrm{z} 344[\mathrm{M}-\mathrm{AcOH}]^{+}$in the EIMS of 2. ${ }^{7}$ Further analysis of the spectral data of these two compounds revealed that the only difference was the presence of an additional hydroxyl group in $\mathbf{2}$ instead of one of the ketone groups in 1. The ol efinic proton signal at $\delta 5.12$ correlated with the C-5, C-4, and C-7 in the HMBC spectrum was assigned to $\mathrm{H}-6$, indicating the presence of a double bond between $\mathrm{C}-6$ and $\mathrm{C}-7$. The proton signal at $\delta 4.46$ was allocated to the oxygenated tertiary carbon at 68.4 bearing a hydroxyl group from the HMQC spectrum and was then assigned to $\mathrm{H}-8$ on the basis of correlations with C-6 and C-9, suggesting the hydroxyl group was attached to C-8. A NOESY spectrum was performed to determine the stereochemistry of the hydroxyl group. The $\mathrm{H}-8$ proton signal at $\delta 4.64(\mathrm{br} \mathrm{d}, \mathrm{J}=10.9 \mathrm{~Hz}$ ) correlated with $\mathrm{H}-5 \beta$ at $\delta 2.34(\mathrm{dd}, \mathrm{J}=14.0,4.8 \mathrm{~Hz})$ and $\mathrm{H}-9 \beta$ at $\delta$ $1.85(\mathrm{~m})$ and was assigned a $\beta$-configuration, indicating that $\mathrm{OH}-8$ was in the $\alpha$-orientation. In the NOESY spectrum, $\mathrm{H}-5 \beta$ could be differentiated from $\mathrm{H}-5 \alpha$ by the correlations with $\mathrm{H}-3 \alpha$ since only $\mathrm{H}-5 \alpha$ was observed to correlate with the latter proton signal. The H-8 $\beta$ was also correlated with the $\mathrm{H}-9 \beta$ resonance at $\delta 2.26$, which could be distinguished from the $\mathrm{H}-9 \alpha$ signal at $\delta 1.85$, because $\mathrm{H}-9 \beta$ was more deshielded by the C-20 lactone carbonyl and shifted downfield. Accordingly, the structure of pseudolaric acid G was determined as 2.

Pseudolaric acid $\mathrm{H}(3)$, with a mol ecular formula $\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{O}_{6}$ as determined by HREIMS ([M] ${ }^{+} \mathrm{m} / \mathrm{z}$ 386.1712, calcd 386.1729), was isolated as an amorphous powder. The ${ }^{1} \mathrm{H}$ NMR (Table 1) and ${ }^{13} \mathrm{C}$ NMR (Table 2) data showed the presence of four methyls, three methylenes, seven methines, and eight quaternary carbons. From the ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, and HMQC spectra, the occurrence of two disubstituted double bonds and two trisubstituted double bonds, three carbonyls, and two oxygenated quaternary carbons was evident. The absence of a molecular ion and the presence of a characteristic weak ion at $\mathrm{m} / \mathrm{z} 326$ [M $\mathrm{AcOH}]^{+}$in the EIMS suggested that 3 is a pseudolaric acid A anal ogue. ${ }^{7}$ The ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectral data of $\mathbf{3}$ indicated that this compound possesses the same side chain as $\mathbf{1}$ and 2. Analysis of spectral data of compound $\mathbf{3}$ showed the presence of one more disubstituted double bond and the absence of a hydroxyl group at C-8 on comparison with compound 2. The presence of a double bond between C-8 and C-9 was then determined on the grounds of HMBC NMR spectral correlations. The ol efinic signal at $\delta 5.70$ was assigned to $\mathrm{H}-9$ on the basis of the correlations with the quaternary carbon signal at $\delta 60.3$ (C-10) and the carbonyl signal at $\delta 173.2$ (C-20). The olefinic proton signal at $\delta 5.74$ was consequently assigned to $\mathrm{H}-8$ from the correlations between itself and two olefinic carbon signals at $\delta 135.0$ (C-7) and 128.7 (C-9), as well as the quaternary carbon signal at $\delta 84.6$ (C-10). The assignments for $\mathrm{H}-8$ and $\mathrm{H}-9$ were confirmed by their strong correlations between the analogous two protons in the HM QC spectrum. The occurrence of a double bond between C-6 and C-7 was also
revealed by the HMBC spectrum, in which one olefinic signal at $\delta 5.38$ was assigned to $\mathrm{H}-6$ judging from the correlation with the oxygenated quaternary carbon signal of C-4 at $\delta 87.2$ and the carbon signal of C-5 at $\delta 33.6$. The structure of pseudolaric acid H was thus elucidated as 3.

2',3'-Dihydroxy-1'-propoxypseudolarate B (4) was assigned the molecular formula $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{O}_{10}$ by HREIMS ( $\mathrm{m} / \mathrm{z}$ $488.2051\left[\mathrm{M}-\mathrm{H}_{2} \mathrm{O}\right]^{+}$, calcd 488.2046) and ESIMS (m/z 529 $[\mathrm{M}+\mathrm{Na}]^{+}$). The ${ }^{1} \mathrm{H}$ NMR (Table 1) and ${ }^{13} \mathrm{C}$ NMR (Table 2) data showed the presence of four methyls, seven methylenes, six methines, and nine quaternary carbons. The ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, and MS data were used to infer that 4 is a derivative of pseudolaric acid $B .{ }^{1 a, b, 5}$ Analysis of the ${ }^{1} \mathrm{H}$ NMR, ${ }^{13}$ C NMR, and EIMS data of 4 indi cated the presence of a 2',3'-di hydroxy-1'-propoxyl group, which was supported by correlations between $\mathrm{H}_{2}$-1 $^{\prime}(\delta 4.20,4.11), \mathrm{H}-2^{\prime}(\delta 3.87)$, and $\mathrm{H}_{2}-3^{\prime}(\delta 3.56)$ in the HMQC spectrum. In the HMBC spectrum, the oxygenated methylene proton signals at $\delta$ 4.20 and 4.11 assigned for the two $\mathrm{H}_{2}-1^{\prime}$ protons were correlated with the carbonyl signal of $\mathrm{C}-18$ at $\delta 168.5$, indicating that the $2^{\prime}, 3^{\prime}$-dihydroxy-1'-propoxyl group is attached to C -18 in the side chain to form a glyceroyl ester of pseudolaric acid B. A characteristic ion at $\mathrm{m} / \mathrm{z} 372$ [M $\left.\mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}_{2}-\mathrm{AcOH}\right]^{+}$in the EIMS supported this conclusion. The structure of compound 4 was therefore elucidated as 2',3'-dihydroxy-1'-propoxypseudolarate B.

6'-O-A cetyl pseudol aric acid B-O- $\beta$-D-glucopyranoside (5) was assigned the molecular formula $\mathrm{C}_{31} \mathrm{H}_{40} \mathrm{O}_{14}$ by HREIMS (m/z 414.1662 [M - AcOH - 162 (glc)] ${ }^{+}$corresponding to $\mathrm{C}_{23} \mathrm{H}_{26} \mathrm{O}_{7}$, calcd 414.1679) and ESIMS (m/z $\left.659[\mathrm{M}+\mathrm{Na}]^{+}\right)$. The ${ }^{1} \mathrm{H}$ NMR (Table 1) and ${ }^{13} \mathrm{C}$ NMR (Table 2) data showed the presence of five methyls, eight methylenes, 11 methines, and 10 quaternary carbons. Comparison of the spectral data of compound $\mathbf{5}$ with those of pseudolaric acid B-O- $\beta$-D-glucoside ${ }^{5}$ inferred that the structures of the two compounds were very similar. However, the proton signals at $\delta 4.94(\mathrm{dd}, \mathrm{J}=16.9,11.9 \mathrm{~Hz})$ and $4.78(\mathrm{dd}, \mathrm{J}=16.9$, 12.1 Hz ) for the two protons of $\mathrm{H}_{2}-6^{\prime}$ were shifted downfield about 1 ppm by the effect of acylation, clearly indicating that one acetoxyl group was linked to the C-6' position. The structure of compound 5 was hence assigned as 6'-Oacetylpseudolaric acid B-O- $\beta$-D-glucopyranoside.

Pseudolaric acids $A, B$, and $C_{1}$, deacetylpseudolaric acid $C_{2}$, deacetylpseudolaric acid A, methyl pseudolarate A, methyl pseudolarate B , pseudolaric acid $\mathrm{A}-\mathrm{O}-\beta-\mathrm{D}-\mathrm{gluco}$ pyranoside, and pseudolaric acid B-O- $\beta$-D-glucopyranoside were identified on the basis of spectral data ${ }^{1,2 b, 6}$ and comparison with authentic samples (co-TLC).

## Experimental Section

General Experimental Procedures. Optical rotations were determined on a Perkin-Elmer 341 polarimeter. UV spectra were measured on a Shimadizu UV-210A spectrometer. IR spectra were recorded on a Perkin-Elmer 577 spectrometer with KBr disks. NMR spectra were measured on a Bruker AM-400 spectrometer with TMS as internal standard. EIMS ( 70 eV ) and ESIMS were carried out on a Finnigan MAT 95 mass spectrometer and a Finnigan LCQDECA instrument, respectively. All solvents used were of analytical grade (Shanghai Chemical Plant, Shanghai, People's Republic of China). Silica gel (200-300 mesh) was used for column chromatography, and a precoated silica gel GF254 plate (Qingdao Haiyang Chemical Plant, Qingdao, People's Republic of China) was used for TLC. C 18 reversed-phase silica gel (150-200 mesh, Merck) and MCl gel (CHP20P, 75-150 $\mu \mathrm{m}$, Mitsubishi Chemical Industries Ltd.) were also used for column chromatography.

Plant Material. The root bark of Pseudolarix kaempferi (Pinaceae) was collected in J iangxi Province of the People's

Republic of China and identified by Professor Zeng-Tao Wang of Shanghai ChineseTraditional Medical University, where a voucher specimen has been deposited in the Herbarium (accession number TJ P-1999-1Y).

Extraction and Isolation. The dried root bark ( 10 kg ) of P. kaempferi was ground and percolated with $95 \%$ ethanol. After filtration and removal of the solvent, the ethanol extract was dissolved in 2 L of $5 \% \mathrm{NaHCO}_{3}$ solution to make a suspension and immediately extracted with ethyl acetate to give a neutral EtOAc-soluble fraction ( 40 g ). The aqueous solution was then treated with $5 \% \mathrm{HCl}$ solution to about pH 6 and extracted with EtOAc again to afford an acidic EtOAcsol uble fraction ( 78 g ). The acidic EtOAc-soluble fraction was subjected to silica gel column chromatography using a gradient solvent system of petroleum-EtOAc (4:1 to 2:1), and then chloroform-methanol ( $10: 1$ to 0:1), and seven major fractions $1-7$ were collected. The first of these fractions was then chromatographed sequentially over a MCl gel CHP 20P column ( MeOH -water, 6:4 to 10:0), a silica gel column (petroleum-EtOAc, 3:1 to 2:1), and a reversed-phase silica gel column ( MeOH -water, 7:3) to afford pseudolaric acid F (1) $(21.3 \mathrm{mg})$, pseudol aric acid G (2) (3.3 mg), pseudolaric acid H (3) $(4.6 \mathrm{mg})$, methyl pseudolarate $\mathrm{A}(10 \mathrm{mg})$, and methyl pseudolarate B ( 690 mg ). Fraction 2 was recrystallized from petrol eum ether-acetone (4:1) to afford pseudolaric adid A (630 mg ). The filtrate was condensed to give a yellow residue, which was then column chromatographed over silica gel eluted with petroleum-EtOAc (2:1) to afford deacetylpseudolaric acid A ( 3 mg ). Fraction 3 was recrystallized from petroleum etheractone ( $4: 1$ ) to yield pseudolaric acid B ( 9.6 g ), and the filtrate was subjected to reversed-phase silica gel column chromatography eluted with MeOH -water (5:5) to afford pseudolaric acids $\mathrm{C}_{1}(53.8 \mathrm{mg})$ and deacetylpseudolaric acid $\mathrm{C}_{2}(25 \mathrm{mg})$. Fraction 4 was chromatographed sequentially over a silica gel column eluted with chloroform-methanol ( $50: 1$ to 10:1), a MCl gel CHP 20P column eluted with methanol-water (8:2), and a reversed-phase $\mathrm{C}_{18}$ silica gel column eluted with methanolwater (6:4) to provide compounds $4(4.7 \mathrm{mg})$ and $5(67 \mathrm{mg})$. Fraction 5 was subjected to silica gel column chromatography eluted with a mixture of chloroform-methanol (10:1) to yield a major compound, pseudol aric acid B-O- $\beta$-D-glucopyranoside ( 9.54 g ), and a minor component. The minor compound, which contained a small amount of impurity, was then subjected to passage over a reversed-phase $\mathrm{C}_{18}$ silica gel column eluted with methanol-water (5:5) to afford pseudolaric acid A-O- $\beta$-Dglucopyranoside ( 270 mg ).

Pseudolaric acid F (1): white amorphous powder; $[\alpha]^{20}{ }_{D}$ $+25.1^{\circ}$ (c 0.93, Me2CO); UV (MeCN ) $\lambda_{\text {max }}(\log \epsilon) 258$ (4.52) nm; IR (KBr) $v_{\max } 3446,2926,1741,1680,1640,1613,1373,1236$, 1171, 1049, 982, 945, $754 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR, see Table 1; ${ }^{13} \mathrm{C}$ NMR, see Table 2; EIMS m/z 384 (1), 368 (5), 342 (7), 298 (30), 283 (10), 237 (10), 225 (17), 199 (20), 188 (18), 160 (19), 149 (26), 117 (19), 91 (35), 77 (21), 60 (100), 57 (32); HREIMS m/z [ $\left.\mathrm{M}-\mathrm{H}_{2} \mathrm{O}\right]^{+} 384.1571\left(\mathrm{C}_{22} \mathrm{H}_{24} \mathrm{O}_{6}\right.$, calcd 384.1573).

Pseudolaric acid G (2): white amorphous powder; $[\alpha]^{20}{ }_{D}$ $-17.4^{\circ}$ (c 0.71, Me2CO); UV (MeCN) $\lambda_{\text {max }}(\log \epsilon) 259$ (4.42) nm; IR (KBr) $v_{\text {max }} 3446,2970,1739,1720,1641,1610,1448,1371$, 1244, 1173, 1057, 1026, 982, 955, $754 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR, see Table 1; ${ }^{13} \mathrm{C}$ NMR, see Table 2; ESIMS m/z 427 [M + Na] ${ }^{+}, 831$ [2M + Na] ${ }^{+}$; EIMS m/z 386 (1), 368 (4), 344 (19), 326 (58), 298 (18),

281 (60), 265 (30), 245 (37), 211 (25), 199 (70), 161 (71), 147 (82), 133 (60), 105 (70), 91 (100), 77 (55), 55 (50); HREIMS $\mathrm{m} / \mathrm{z}\left[\mathrm{M}-\mathrm{H}_{2} \mathrm{O}\right]^{+} 386.1764\left(\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{O}_{6}\right.$, calcd 386.1729).

Pseudolaric acid H (3): gum; $[\alpha]^{20} \mathrm{D}+11.5^{\circ}$ (c 0.56, Me2CO); UV (MeCN) $\lambda_{\text {max }}(\log \epsilon) 257$ (4.67) nm; IR (KBr) $\nu_{\max } 3446$, 2976, 1740, 1641, 1612, 1439, 1369, 1240, 1157, 1043, 981 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR, see Table 1 ; ${ }^{13} \mathrm{C}$ NMR, see Table 2; ESIMS $\mathrm{m} / \mathrm{z} 409[\mathrm{M}+\mathrm{Na}]^{+}, 795[2 \mathrm{M}+\mathrm{Na}]^{+}$; EIMS m/z 386 (1), 327 (3), 326 (18), 309 (4), 308 (20), 298 (4), 290 (3), 283 (5), 282 (24), 281 (21), 267 (25), 265 (5), 221 (4), 195 (8), 183 (47), 172 (16), 168 (11), 145 (11), 144 (45), 143 (19), 129 (13), 128 (12), 105 (11), 91 (16), 68 (100); HREIMS m/z [M ]+ 386.1712 ( $\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{O}_{6}$, calcd 386.1729).
2,3'-Dihydroxy-1'-propoxypseudolarate B (4): white amorphous powder; $[\alpha]^{20}{ }_{D}-18.3^{\circ}$ (c 0.46, Me2CO); UV (MeCN) $\lambda_{\text {max }}(\log \epsilon) 261(4.45) \mathrm{nm} ;$ IR (KBr) $v_{\text {max }} 3435,2953,1740,1707$, 1632, 1608, 1446, 1383, 1232, 1165, $752 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR, see Table 1; ${ }^{13} \mathrm{C}$ NMR, see Table 2; ESIMS m/z 529 [M + Na] ${ }^{+}$, 1035 [2M + Na]+; EIMS m/z 488 (2), 447 (2), 414 (10), 372 (5), 355 (10), 354 (33), 342 (4), 334 (11), 328 (16), 316 (19), 310 (14), 296 (12), 295 (10), 284 (15), 274 (14), 260 (16), 257 (13), 242 (38), 224 (36), 191 (45), 137 (37), 128 (30), 131 (100), 117 (32), 115 (31), 109 (36), 105 (44), 91 (83), 61 (79); HREIMS $\mathrm{m} / \mathrm{z}\left[\mathrm{M}-\mathrm{H}_{2} \mathrm{O}\right]^{+} 488.2051\left(\mathrm{C}_{26} \mathrm{H}_{32} \mathrm{O}_{9}\right.$, calcd 488.2046).

6'O-Acetylpseudolaric acid B-O- $\beta$-D-glucopyranoside (5): white amorphous powder; $[\alpha]^{20}{ }^{\mathrm{D}}-7.2^{\circ}$ (c $0.77, \mathrm{Me} \mathrm{CO}$ ); UV (MeCN) $\lambda_{\text {max }}(\log \epsilon) 265$ (4.47), 205 (4.59) nm; IR (KBr) $v_{\max } 3446,2953,1736,1717,1716,1643,1610,1444,1371$, 1279, 1232, 1074, $750 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR, see Table $1 ;{ }^{13} \mathrm{C}$ NMR, see Table 2; ESIMS m/z 659 [M + Na]+; EIMS m/z 414 (10), 372 (18), 342 (19), 328 (16), 296 (21), 260 (100), 224 (45), 191 (39), 169 (17), 131 (38), 91 (27), 77 (15); HREIMS m/z [M -$\mathrm{AcOH}-162]^{+} 414.1662\left(\mathrm{C}_{23} \mathrm{H}_{26} \mathrm{O}_{7}\right.$, calcd 414.1679).

Acknowledgment. Financial support of the National Scientific Foundation (for J.-M.Y., 30025044) and the Shanghai Municipal Scientific Foundation for Fundamental Research (for J.-M.Y., 00J C14053) is gratefully acknowledged. We thank Zeng-Tiao Wang of Shanghai Chinese Traditional Medical University for identification of the plant material.

## References and Notes

(1) (a) Li, Z. L.; Pan, D. J.; Hu, C. Q.; Wu, Q. L.; Yang, S. S.; Xue, G. Y. Huaxuexuebao 1982, 40, 447-456. (b) Zhou, B. N.; Ying, B. P.; Song, G. Q.; Chen, Z. X.; Han, J .; Yan, Y. F. Planta Med. 1983, 47, 35-38. (c) Li, Z. L.; Chen, K.; Pan, D. J.; Xue, G. Y. Huaxue Xuebao 1985, 43, 786-788. (d) Li, Z. L.; Chen, K.; Pan, D. J.; Xue, G. Y. Huaxue Xuebao 1989, 47, 258-261.
(2) (a) Chen, K.; Li, Z. L.; Pan, D. J.; Xue, G. Y. Huaxue Xuebao 1990, 48, 591-595. (b) Yang, S.; Yue, J. Bi oorg. Med. Chem. Lett. 2001, 11, 3119-3122.
(3) (a) Chen, G.; Li, Z.; Pan, D.; Tang, C.; He, X.; Xu, G.; Chen, K.; Lee, K. J . Nat. Prod. 1993, 56, 1114-1122. (b) Chen, G.; Li, Z.; Chen, K.; Tang, C.; He, X.; Pan, D.; Hu, C.; McPhail, D. R.; McPhail, A. T.; Lee, K. Chem. Commun. 1990, 1113-1114.
(4) Pan, D.; Li, Z.; Hu, C.; Chen, K.; Chang, J .; Lee, K. Planta Med. 1990, 56, 383-385.
(5) Li, E.; Clark, A. M.; Hufford, C. D. J . Nat. Prod. 1995, 58, 57-67.
(6) Li, X.; El Sohly, H. N.; Nimrod, A. C.; Clark, A. M.J . Nat. Prod. 1999, 62, 767-769.
(7) Huang, Z.; Liu, B.; Yin, B.; Yu, Q.; Han, M. Huaxue Xuebao 1984, 42, 886-892.

NP0200010


[^0]:    * To whom correspondence should be addressed. Tel: 86-21-64311833. Fax: 86-21-64370269. E-mail: jmyue@mail.shcnc.ac.cn.

[^1]:    ${ }^{\text {a }}$ Data expressed in ppm, with J values in Hz . ${ }^{\text {b }}$ Compounds $\mathbf{1 - 4}$ were measured in acetone- $\mathrm{d}_{6}$, and compound $\mathbf{5}$ was recorded in pyridine$\mathrm{d}_{5}$.

[^2]:    a Data expressed in ppm. ${ }^{\text {b }}$ Compounds $\mathbf{1}-\mathbf{4}$ were measured in acetone-d ${ }_{6}$, and compound 5 was recorded in pyridine-d ${ }_{5}$.

